## Stereochemistry - structure in three dimensions

Isomers that have the same constitution but differ in their 3-D arrangement are **stereoisomers**.

cis and trans isomers are one type of stereoisomers.

now look at another type of stereoisomerism ...

Every object has a mirror image, but not all objects are superimposable on their mirror image.

Consider a pair of objects ... a pair of hands

These objects are not identical; they are mirror images of each other. The mirror images are not superimposable upon each other.

An object that is not superimposable on its mirror image is chiral. chiral means "handedness"

The opposite of chiral is **achiral**.

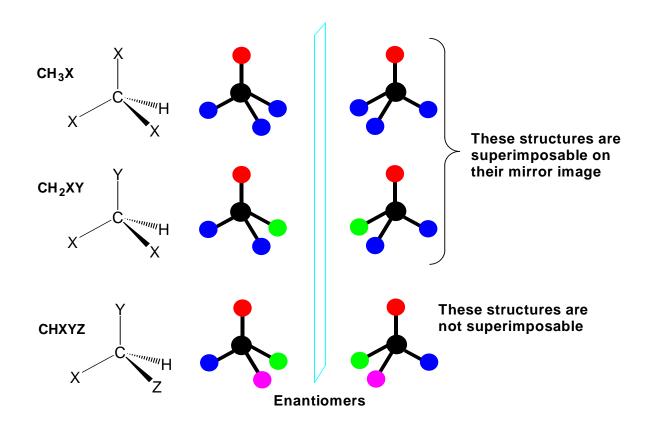
- A sugar cube is achiral because it is superimposable on its mirror image.
- Hands are chiral, gloves are chiral but socks are generally achiral.

Many molecules are chiral.

Chirality most often occurs in molecules that have 4 different groups attached to carbon.

Consider, CWXYZ and its mirror image; they are non superimposable.

Objects (or molecules) that are non superimposable on their mirror image are **enantiomers**.



Molecules with 4 different groups on carbon are chiral.

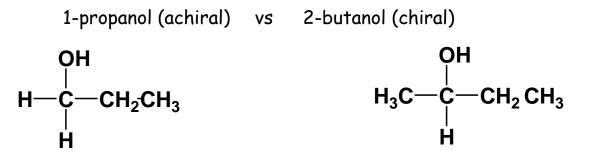
This type of carbon atom is called a chirality center.

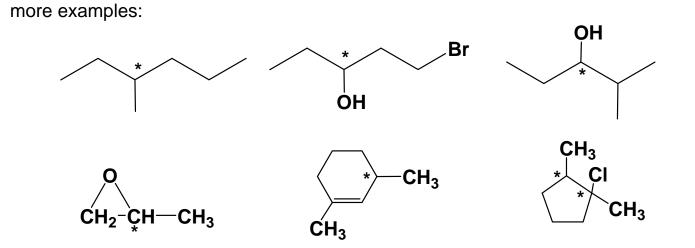
Similar terms: asymmetric center, stereogenic center or chiral center.

## Test for chirality:

## Look for 4 different groups attached to carbon

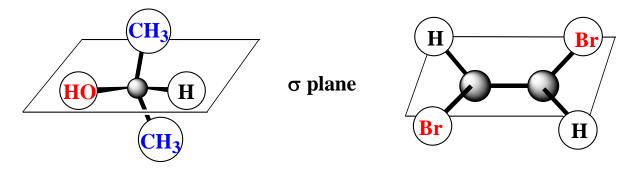
Look at more examples: 4 different groups on carbon





## Test for achiral structures:

- Look for a plane of symmetry, also called a  $\sigma$  plane
- Or center of symmetry
- > Plane of symmetry



Center of symmetry- Chirality due to this type of situation is not as common as applying the symmetry plane test or identifying 4 different groups on carbon. <u>Students do not need to be able to distinguish</u> <u>centers of symmetry in molecules.</u>

